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Patents ADP number <i>(if you know it)</i>	7805310001		
If the applicant is a corporate body, give the country/state of its incorporation	U.S.A. State of Delaware		
4. Title of the invention	ABRASIVE ARTICLES		
5. Name of your agent <i>(if you have one)</i>	LLOYD WISE		
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6th November, 2002

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ABRASIVE ARTICLES

The present invention relates to abrasive articles. In particular the present
5 invention relates to abrasive articles comprising a UV-cured binder and filler.

Abrasive articles typically comprise a plurality of abrasive particles and a
binder. There are a number of different types of abrasive articles on the
market. These include coated abrasive products, bonded abrasive products
10 and nonwoven abrasive products.

Coated abrasive products generally include a backing, abrasive particles, and
at least one binder to hold the abrasive particles onto a major surface of the
backing. The abrasive layer can be, for example, a single layer (e.g., a slurry
15 layer) or multiple layers (e.g., make and size layers). The slurry layer may be
applied as a slurry of abrasive particles in a binder precursor that is
subsequently cured to form the binder. Such slurries of abrasive particles in a
binder precursor and techniques for applying them are well known in the
abrasive art. Make and size layers and methods for applying them are also
20 well known in the abrasive art. In addition to adhering the abrasive particles
to the backing, the make coat may also serve to seal the backing. The
backing may be any suitable material including, but not limited to, cloth,
polymeric film, fibre, woven fabric, nonwoven web, paper, or combinations
thereof, or treated versions thereof. The abrasive particles can be present in
25 one or more layers of the coated abrasive product.

Bonded abrasive products typically include a shaped mass of abrasive
particles held together by an organic, metallic, or vitrified binder. Such
shaped mass can be, for example, in the form of a wheel, such as a grinding
30 wheel or cutoff wheel. The shaped mass can also be in the form, for
example, of a honing stone, segment, mounted point, disc (e.g. double disc
grinder) or other conventional bonded abrasive shape.

Nonwoven abrasive products typically include an open porous lofty polymer filament structure having abrasive particles distributed throughout the structure and adherently bonded therein by an organic binder. Examples of filaments include, but are not limited to, polyester fibers, polyamide fibers, and polyaramid fibers.

Abrasive articles typically include at least one binder (e.g. in make, size, and/or slurry layers of coated abrasive articles, or coated on a fiber web of nonwoven abrasive articles). Typically, binder(s) is/are formed by curing (e.g. thermally, or by using electromagnetic or particulate radiation) binder precursor(s). Useful binders and binder precursors may be inorganic or organic. Useful binder precursors include thermally curable resins and radiation curable resins, which may be cured, for example, thermally and/or by exposure to radiation. Exemplary organic binder precursors include glue, phenolic resin, aminoplast resin, urea-formaldehyde resin, melamine-formaldehyde resin, urethane resin, (e.g. an aminoplast resin having pendant α,β -unsaturated groups, acrylated urethane, acrylated epoxy, acrylated isocyanurate), acrylic resin, epoxy resin (including bis-maleimide and fluorene-modified epoxy resins), isocyanurate resin, as well as mixtures thereof.

Binders used to produce abrasive articles often contain fillers. Fillers are typically organic or inorganic particulates dispersed within the resin and may, for example, modify either the binder precursor or the properties of the cured binder, or both, and/or may simply, for example, be used to reduce cost. In coated abrasives, the fillers may be present, for example, to block pores and passages within the backing to reduce its porosity and provide a surface to which the maker coat will bond effectively. The addition of a filler, at least up to a certain extent, typically increases the hardness and toughness of the cured binder. Inorganic particulate filler commonly has an average particle size ranging from about 1 micrometer to about 100 micrometers, more preferably from about 5 to about 50 micrometers, and sometimes even from about 10 to about 25 micrometers. Depending on the ultimate use of the abrasive article the filler typically has a specific gravity in the range of 1.5 to

4.5, and an average particle size of the filler will preferably be less than the average particle size of the abrasive particles.

Examples of useful fillers include: metal carbonates such as calcium carbonate (in the form of chalk, calcite, marl, travertine, marble or limestone), calcium magnesium carbonate, sodium carbonate, and magnesium carbonate; silicas such as quartz, glass beads, glass bubbles and glass fibers; silicates such as talc, clays, feldspar, mica, calcium silicate, calcium metasilicate, sodium aluminosilicate, and sodium silicate; metal sulfates such as calcium sulfate, barium sulfate, sodium sulfate, aluminium sodium sulfate, and aluminium sulfate; gypsum; vermiculite; wood flour; alumina trihydrate; carbon black; metal oxides such as calcium oxide (lime), aluminium oxide, titanium dioxide, alumina hydrate, alumina monohydrate; and metal sulfites such as calcium sulfite.

The use of UV-curable coating compositions in the preparation of coated abrasives is known. One of the problems associated with the use of UV-curable coating compositions in coated abrasives is that heavy filler loadings cause a shadowing effect and are inherently difficult to cure because the composition behind particles of filler tend to prevent penetration of UV-radiation throughout the depth of the coating. The problem is exacerbated in make coatings since the presence of the abrasive particles embedded in the partially cured coating also contributes to the shadowing effect.

WO97/36713, published October 9, 1997, reports a coating composition suitable for the production of coated abrasives comprising a UV-polymerisable formulation and an alumina trihydrate filler that is substantially transparent to UV-light, wherein the amount of alumina trihydrate is said to be from 25 to 50% by volume of the coating composition.

The present invention provides abrasive articles, particularly coated abrasives comprising a UV-cured formulation and filler.

According to the present invention there is provided an abrasive article comprising abrasive particles and binder material, wherein the binder material comprises a UV- cured formulation and aluminosilicate ceramic microspheres substantially transparent to UV- radiation, wherein the aluminosilicate ceramic microspheres are present in a range from 20 to 80 percent by weight, based on the combined weight of the cured formulation and the aluminosilicate ceramic microspheres, wherein the aluminosilicate ceramic microspheres have an average particle size in a range from 1 micrometer to 40 micrometers.

In a preferred embodiment the abrasive article is in the form of a coated abrasive comprising abrasive particles which are supported on and adherently bonded to at least one major surface of a backing sheet by a make coating of a first resinous material and a size coating of a second resinous material, wherein at least some of said first or second resinous materials comprises said UV-cured binder and filler.

It has been found that solid microspheres which are transparent to UV- radiation are particularly suitable for coatable compositions comprising a UV- polymerisable binder system. Typically, the microspheres may be incorporated in the compositions in high levels (e.g. up to 80 weight %) while still providing acceptable viscosities for coating. The compositions may be readily cured by exposure to UV-light which is able to penetrate deeply into the coated compositions. In most UV-curable binder systems the flexural modulus increases with higher levels of microspheres. The coatable compositions can be used, for example, for the production of abrasive articles (e.g., coated abrasives). Advantages of curable compositions according to the present invention include lower raw material cost as the microsphere filled UV-curable formulation is lower in cost compared with an unfilled UV curable system. The use of other conventional fillers, which could reduce cost, prevent the passage of UV radiation through the material, increase the viscosity markedly, and reduce or entirely prevent curing. Reduced curing results in a reduction in physical properties such as modulus and toughness, and therefore render the material unsatisfactory in an abrasive formulation.

The microspheres used in the present invention are transparent to light in the range 25 nanometers to 450 nanometers. In some embodiments, for example, where the coatable composition comprises visible light activated catalysts, the microspheres are transparent to visible light. The microspheres are formed of ceramic and have an average particle size in a range from 1 micrometer to 40 micrometers preferably in a range from 1 micrometer to 20 micrometers. Ceramics comprise all engineering materials or products that are chemically inorganic, including glass, metal oxides, crystalline metal oxides and glass-ceramic metal oxides, except metals and alloys and are usually rendered serviceable through high temperature processing. Suitable microspheres are commercially available, for example, under the trade designation "ZEEOSPHERE" from the 3M Company, St. Paul, MN, USA. In some embodiments, preferred microspheres include those available under the trade designation "W210 ZEEOSPHERES", from the 3M Company, which have an average particle size of about 3 micrometers and a particle size distribution in a range from 1 micrometer to 12 micrometers.

A filler is considered for the purposes of this invention to be transparent to this light if, when a composition containing the UV-polymerisable formulation and 25% by weight of the uniformly distributed filler, is exposed to UV light the depth of cure obtained is at least 50% of the depth attained when the formulation without the filler receives the same amount of UV radiation.

The depth of cure is measured by coating a composition on a surface and exposing under a UV source for a predetermined time. The result is the formation of a thin crust on the surface of the coating. The thickness of this crust is a measure of the relative depth of penetration of the UV radiation.

Typically, for curable compositions according to the present invention used to make coated abrasive articles, the microspheres are present in an amount from 20 to 80% by weight of the composition, preferably from 40 to 60% of the composition.

Compositions according to the present invention may further comprise other fillers in addition to the microspheres. However, since most fillers are not transparent to UV-radiation and the presence of significant amounts of such fillers may deleteriously affect the curing properties of the composition.

5 Surprisingly, it has been found that mica may be advantageously used as a filler in combination with the UV transparent microspheres. While not wanting to be bound by theory, it is believed that although mica is opaque to UV-radiation, it is transparent to visible light and the presence of a visible light activated catalyst in the compositions ensures full cure of the binder system.

10 In some embodiments, the compositions comprise up to at least 22% by weight, higher loadings of mica tend to increase the viscosity of compositions such that they are difficult to coat. Further, in some embodiments, the weight of microspheres in the coatable composition is preferably greater than the weight of mica.

15 The UV-curable component of the compositions of the present invention may comprise such materials known in the art suitable for the production of abrasive articles, wherein the UV curable component is curable on exposure to UV light in the wavelength range 10nm to 400nm, and/or in the presence of
20 a visible light activated catalyst cures on exposure to visible light in the wavelength range 400nm to 700nm.

Suitable UV curable resins typically are resins that polymerise via a free-radical mechanism. They include epoxy-acrylates, aminoplast derivatives
25 having pendant α,β -unsaturated carbonyl groups, ethylenically unsaturated compounds, isocyanurate derivatives having at least one pendant acrylate group, isocyanates having at least one pendant acrylate group, urethane-acrylate, epoxy-novolacs and mixtures thereof.

30 Examples of acrylated urethanes include diacrylate esters of hydroxy terminated isocyanate extended polyesters or polyethers. Acrylated epoxies include, for example, the diacrylate esters of bisphenol derivatives such as bisphenol A epoxy resins. Typical aminoplast derivatives have at least 1.1 pendant α,β -unsaturated carbonyl groups. Suitable ethylenically unsaturated

compounds include monomeric or polymeric compounds that contain atoms of carbon, hydrogen and oxygen, and optionally nitrogen and the halogens.

Oxygen and nitrogen atoms are generally present in ether, ester, urethane, amide or urea groups. Typically isocyanate derivatives have at least one

5 pendant acrylate group, conventionally made, for example, by the reaction of an acrylate monomer or oligomer, including di- and tri-acrylates, with a novolac, epoxy or urethane polymer or oligomer.

Most binder systems curable by UV-radiation require the presence of a

10 photoinitiator to initiate free-radical polymerisation. Non-limiting examples of suitable photoinitiators include, benzophenones, phosphine oxides, nitroso compounds, acryl halides, hydrazones, mercapto compounds, pyrrilium compounds, triacrylimidazoles, benzimidazoles, chloroalkyl triazines, benzoin ethers, benzil ketals, thioxanthenes, camphorquinone, and acetophenone
15 derivatives. Cationic photoinitiators may also be used and examples of such photoinitiators include aryl diazonium, arylsulfonium, arylodonium and ferrocenium salts. Examples of UV-curable binder systems are reported in U.S. Patent Nos. 4,735,632 (Oxman et al.), 4,773,920 (Chasman et al.), 5,152,917 (Pieper et al.), 5,304,223 (Pieper et al.), 5,391,210 (Bilkadi et al.),
20 and 5,667,541 (Klun et al.), the disclosures of which are incorporated herein by reference.

Coatable compositions according to the present invention may further

comprise coupling agents. Coupling agents may function to form a stronger
25 bond between the binder and the inorganic particles of filler and abrasive, or the backing. In some embodiments, preferred coupling agents include organo-functional silanes, for example, vinyl functional and methacrylic functional silanes. Although not wanting to be bound by theory, it is believed the presence of the coupling agent increases the toughness of the binder system.

30

Optionally, coatable compositions according to the present invention may further comprise surfactant and/or other coating aids. The present of surfactant facilitates good filler dispersion and reduces viscosity. Although not wanting to be bound by theory, it is believed that, in some cases, the

presence of surfactant may also increase the flexural modulus of the binder system. Typically, the amount of surfactant is up to about 2% by weight of the composition, generally from 0.3 to 1.2% by weight of composition, although amounts above 2% by weight may also be useful. A wide range of

5 surfactants may be used including those having the trade designations "LICA 09", "KRTTS", "LICA 385", "LICA N709", "K755", "K70PPR" (available from Kenrich Petrochemicals, Inc., Bayonne, NJ, USA), "SILANE GF80" (available from Wacker-Chemie GmbH, Munich, Germany), "BYK 980", "BYK 9010" and "BYK 985" (Byk-Chemie, Wesel, Germany).

10

The backing can be any number of various materials conventionally used as backings in the manufacture of coated abrasives, such as paper, cloth, film, vulcanised fibre, woven and nonwoven materials, and the like, or a combination of two or more of these materials or treated versions thereof.

15

The choice of backing material may depend, for example, on the intended application of the abrasive article. Typically, the strength of the backing should be sufficient to resist tearing or other damage in use, and the thickness and smoothness of the backing should allow achievement of the product thickness and smoothness desired for the intended application. Further, the

20 adhesion of the inventive coatable composition or other binder to the backing typically should also be sufficient to prevent significant shelling of individual abrasive particles or the abrasive coating during normal use. In some applications, it is preferable that the backing be waterproof. The thickness of the backing should be sufficient to provide the strength desired for the

25 intended application; nevertheless, it should not be so thick as to affect the desired flexibility in the coated abrasive product. For lapping coated abrasives, one preferred backing is polymeric film, such as polyester film, and that the film be primed with a material, such as ethylene acrylic acid copolymer, to promote adhesion of the inventive slurry or dispersion and

30 resulting abrasive composite to the film. In some embodiments of the coated abrasive article it may be preferable to utilise a backing transparent to UV/visible radiation.

Embodiments of coatable composition according to the present invention are useful, for example, for the preparation of coated abrasives. For example, the compositions may be used as the make and/or size coat of the coated abrasive article. In some embodiments, the size coat of the coated abrasive article is preferably made using the composition of the present invention.

In the case of a woven backing, it is sometimes preferable, for example, to fill the interstices of the backing with at least one coating before the application of an inventive coatable composition. Coatings used for this purpose are called saturant, back or presize coatings, as previously described, depending on how and to what surface of the backing the coating is applied.

In some embodiments, the backing may comprise a laminate of backings made by laminating two or more plies of either similar or dissimilar backing materials. For example, the backing can be laminated to a stiffer, more rigid substrate, such as a metal plate, to produce a coated abrasive article having an abrasive coating supported on a rigid substrate.

The major surface of the backing opposite the abrasive coating may, in some embodiments, include a pressure-sensitive adhesive or one member of a hook and loop type attachment system so that the abrasive article can be secured to a back-up pad. Examples of pressure-sensitive adhesives suitable for this purpose include rubber-based adhesives, acrylate-based adhesives, and silicone-based adhesives.

Suitable abrasive particles may be selected from those commonly used in the abrasive art, however, the abrasive particles (size and composition) will be chosen, for example, with the application of the abrasive article in mind. In choosing an appropriate abrasive particle, characteristics such as light absorption, hardness, compatibility with the intended workpiece, particle size, reactivity with the workpiece, as well as heat conductivity may also be considered.

The composition of abrasive particles useful in the present invention can be divided into two classes: natural abrasives and manufactured abrasives.

Examples of useful natural abrasives include: diamond, corundum, emery, garnet (off-red colour), buhrstone, chert, quartz, sandstone, chalcedony, flint, quartzite, silica, feldspar, pumice and talc. Examples of manufactured abrasives include: boron carbide, cubic boron nitride, fused alumina, ceramic
5 aluminium oxide, heat treated aluminium oxide (both brown and dark grey), alumina zirconia, glass, silicon carbide (preferably green, although small amounts of black may be tolerated), iron oxides, tantalum carbide, cerium oxide, tin oxide, titanium carbide, synthetic diamond, manganese dioxide, zirconium oxide and silicon nitride.

10

In some embodiments abrasive particles useful in the present invention typically have a particle size ranging from about 0.1 micrometer to about 1500 micrometers, more typically ranging from about 0.1 micrometer to about 1300 micrometers. Typically, the abrasive particles used in the present invention
15 have a Moh's hardness of at least 8, more typically above 9; however, abrasive particles having a Moh's hardness of less than 8 may be used.

15

The term "abrasive particle" includes agglomerates of individual abrasive particles. An abrasive agglomerate is formed when a plurality of abrasive
20 particles are bonded together with a binder to form a larger abrasive particle which may have a specific particulate structure. The plurality of particles which form the abrasive agglomerate may comprise more than one type of abrasive particle, and the binder used may be the same as or different from the binders used to bind the agglomerate to a backing.

25

Precisely shaped abrasive particles may also be employed. These precisely shaped abrasive particles may be produced essentially by coating a structured tool with a slurry of abrasive particles and binder precursor, curing the binder precursor, and detaching the resulting composites from the tool by
30 exposing the tool to an ultrasonic horn, as reported in U.S. Pat. Nos. 5,152,917 (Pieper et al.) and 5,549,962 (Holmes et al.), the disclosure of which are incorporated herein by reference.

30

Advantages and embodiments of this invention are further illustrated by the following Examples, but the particular materials and amounts thereof recited in these examples, as well as other conditions and details, should not be construed to unduly limit this invention. All parts and percentages are by weight unless otherwise indicated. In the Tables the amounts of the components of the formulations are in parts by weight unless otherwise indicated.

In the Examples the following materials were used:

"P.80" refers to blue aluminium oxide mineral, obtained under the trade designation "BFRPL" from Treibacher Schleifmittel UK Ltd, The Old Barn, Dunsborough Park, Ripley, Woking, Surrey, GU23 6AL, UK.

Trimethylol propane triacrylate obtained under the trade designation "TMPTA" from UCB Radcure Specialties, Smyrna, GA, USA.

Vinylester resin obtained under the trade designation "GENOMER 2258" from Rahn AG, Dorfstrasse 120, CH-8050, Zurich, Switzerland.

Hexandiol diacrylate obtained under the trade designation "GENOMER 1223" from Rahn AG

Ethoxylated trimethylolpropane triacrylate obtained under the trade designation "GENOMER 1343" from Rahn AG

Vinylester resin obtained under the trade designation "GENOMER 2258" from Rahn AG

Wetting agent obtained under the trade designation "GENORAD 10" from Rahn AG

Cycloaliphatic epoxy resin obtained under the trade designation "UVACURE 1500" from UCB Radcure Specialties.

Cycloaliphatic epoxy resin obtained under the trade designation "ERL 4221" from Dow Chemical Co., Midland, MI, USA.

- 5 Free radical catalyst obtained under the trade designation "DAROCURE 1173" from Ciba Speciality Chemicals, Basel, Switzerland.

Cationic catalyst obtained under the trade designation "CPI6976" from Aceto Corp., Lake Success, NY, USA.

- 10 Cationic catalyst obtained under the trade designation "UVI 6974" from Dow Chemical Co.

- 15 Alkali aluminosilicate ceramic microspheres obtained under the trade designation "ZEEOSPHERES W 210", "ZEEOSPHERES W 410", "ZEEOSPHERES W 610" from the 3M Company, St. Paul, MN, USA

Silica-alumina ceramic microspheres, obtained under the trade designation "ZEEOSPHERES G200", from the 3M Company.

- 20 A mica obtained under the trade designation "SX400" from Microfine Minerals Ltd., Derby, U.K.

- 25 A feldspar obtained under the trade designation "MINEX" from North Cape Minerals AS, Rud, Norway.

Wetting agent obtained under the trade designation "BYK W-985" from BYK-Chemie GmbH, Abelstraße 45, P.O. Box 100245, D-46462 Wesel, Germany
A vinyl functional silane.

- 30 Methacrylate functional silanes, obtained, respectively, under the trade designation "GF56" and "GF31" from Wacker-Chemie GmbH, Hauptverwaltung, Hanns-Seidel-Platz 4, 81737 München, Germany.

A methacrylate polymer obtained under the trade designation "PARALOID 2655" from Rohm and Haas (UK) Ltd., Lennig House, 2 Mason's Avenue, Croydon, Surrey CR9 3NB, UK.

- 5 A free-radical catalyst obtained under the trade designation "GENOCURE MBF" from Rahn AG.

A free-radical catalyst obtained under the trade designation "IRGACURE 819" from Ciba Speciality Chemicals, Charter Way, Macclesfield, Cheshire SK10 2NX, UK.

10

Photoinitiators obtained under the trade designation "SPEEDCURE 123", "SPEEDCURE 2263", "SPEEDCURE BEM", "SPEEDCURE DMB", "SPEEDCURE EDB", "SPEEDCURE PBZ" and "SPEEDCURE TPO from Lambson Group Ltd ,103-105, Station Parade, Harrogate, N. Yorkshire, HG1 1HB, UK.

15

Surfactants, obtained under the trade designation "LICA 09", "KRTTS", "38J", "KZ55", "KZOPPR", "NZ09" and "NZ38" from Kenrich Petrochemicals Inc., Bayonne, NJ, USA.

20

Example 1

The following components and amounts (see Table 1, below) were used to make Example 1 samples.

25 **TABLE 1**

Component	Example 1.1	Example 1.2	Example 1.3	Example 1.4
"GENOMER 2258"	3.55	3.55	3.55	3.55
"GENOMER 1343"	33.39	33.39	33.39	33.39
"TMPTA"	18.08	18.08	18.08	18.08
"GENOCURE MBF"	2	2	2	2
"SPEEDCURE PBZ"	1.6	1.6	1.6	1.6
"IRGACURE 819"	0.4	0.4	0.4	0.4

"SPEEDCURE EDB"	1.2	1.2	1.2	1.2
"BYK W-985"	0	0.09	0.23	0.51
"W210-ZEEOSPHERES"	0	14.49	37.85	85.28
"GENORAD 10"	1.15	1.15	1.15	1.15
WEIGHT % "ZEEOSPHERES"	0	19	38	58

The compositions were prepared by first adding the resins (Genomer 2258, Genomer 1343, TMPTA), warming to 60°C, then adding the BYK W-985, then adding the filler (W-210 Zeeospheres), mixing ultrasonically until smooth using a Branson 2210 ultrasonic bath supplied by Worldwide Headquarters, Branson Ultrasonics Corp., 41 Eagle Rd., Danbury, CT 06813, USA, then adding the catalysts (Genocure MBF, Speedcure PBZ, Irgacure 819, Speedcure EDB), then adding the Genorad 10 and stirring until smooth and de-gassed for 10 minutes at 60°C and 5 minutes at 60°C respectively. Four slabs of each mix were made by UV-curing in moulds of dimensions 9mm wide x 75mm long x 1mm thick. A single lamp "D-bulb" UV curing unit obtained under the trade designation "MINICURE" from Primark UV Technology of Primark Limited, A Nordson Company, 816 Leigh Road, Slough, Berkshire, SL1 4BD, U.K. was used, with a belt speed of 10 cm/second, and a power setting of the "D-bulb" of 65%, that gave a UV dose of 480 millijoules/cm² and a temperature of 53°C, as measured by a "POWER PUCK" UV dose meter obtained from DDU ENTERPRISES, 2909 Oregon Court, Suite A-2, Torrance, CA 90503, USA. The slabs were then annealed, with a thermal cure of 24 minutes at 110°C.

Example 1.1 had a low viscosity and seemed more brittle than the other formulations. Example 1.2 was less brittle and shrank less. Example 1.3 produced strips that were not noticeably less brittle than Example 1.2 but showed less shrinkage. Example 1.4 showed less shrinkage than the other formulations.

The slabs were sanded against "P1000 WETOrDRY" obtained under the trade designation 3M 312 from the 3M Company followed by 15micrometers then 7micrometers sandpaper micro-finishing film obtained under the trade designation 3M 268L from the 3M Company, by hand sanding parallel to the edges until smooth (about 10 passes on each edge on each grade of sandpaper). This sanding process removed flaws along the edges that would reduce the physical properties. The resulting slabs were tested in three point bend mode on a flexural modulus testing apparatus, obtained from Instron Corp., Canton, MA, USA, under the trade designation "INSTRON 4301". The results are provided in Table 2, below.

TABLE 2

% "ZEEOSPHERES"	Flexural Modulus, Mpa
0	785.4
19	702.4
38	1025.6
58	1935.4

Example 2

The following components and amounts (see Table 3, below) were used to make Example 2 samples. The procedures for making and testing the Example 2 samples was as described in Example 1.

TABLE 3

Component	Example 2.1	Example 2.2	Example 2.3	Comparative 2.4
"GENOMER 2263"	19.87	19.87	19.87	19.87
"TMPTA"	27.76	27.76	27.76	27.76
"GENOMER 1223"	4.30	4.30	4.30	4.30
"GENOCURE MBF"	0.69	0.69	0.69	0.69

"IRGACURE 819"	0.24	0.24	0.24	0.24
"BYK W-985"	0.32	0.32	0.32	0.32
"SX400"	11.70	11.70	11.70	11.70
"W-210 ZEEOSPHERES"	31.94	-	-	-
"W-410 ZEEOSPHERES"	-	31.94	-	-
"W-610 ZEEOSPHERES"	-	-	31.94	-
"G-200 ZEEOSPHERES"	-	-	-	31.94
"GF56"	0.60	0.60	0.60	0.60
"PARALOID 2655"	2.58	2.58	2.58	2.58

The compositions were prepared by first adding the resins (Genomer 2263, Genomer 1223, TMPTA), warming to 60°C, then adding the BYK W-985, then adding the fillers (Zeeospheres, SX400) and Paraloid 2655, mixing ultrasonically until smooth using a Branson 2210 ultrasonic bath supplied by Worldwide Headquarters, Branson Ultrasonics Corp., 41 Eagle Rd., Danbury, CT 06813, USA, then adding GF56 Silane with stirring and heating to 75°C for 45 minutes, then adding the catalysts (Genocure MBF, Irgacure 819), then adding the Genorad 10 and stirring until smooth.

The viscosity of the uncured liquid was measured at 23 degrees C using a "HAAKE RHEOSTRESS RS75" rheometer Thermo Haake, Thermo Haake HQ, Dieselstrasse 4, Karlsruhe, BW, 76227, Germany with a 3.5cm cone and plate. In addition to the modulus of Example 2 samples, toughness value was also calculated for each sample by taking the area under the Stress/Strain curve (which equates to the energy required to break the sample = Force x Distance) and dividing by the sample volume between the sample supports (Gauge length x Thickness x Width). Further, it was observed that Comparative 2.4 did not cure on the side opposite the UV source.

The results are reported in Table 4, below.

TABLE 4

	Example 2.1	Example 2.2	Example 2.3	Comparative 2.4
Viscosity, mPas	383	1083	945	-
Modulus, Mpa	7335	6375	6480	-
Toughness, kgs-2m ⁻¹	0.070	0.064	0.076	-

5

“ZEEOSPHERES G-200” silica-alumina ceramic microspheres did not transmit UV radiation.

10 “ZEEOSPHERES W-210, W-410, W-610”, alkali aluminosilicate ceramic microspheres, all produce a suitable cure. Although not wanting to be bound by theory, it is believed that the uncured liquid varies in viscosity. due to the different microsphere diameter. Further, it is believed that, as shown in Table 5 below, the larger the microsphere diameter, the higher the viscosity. This finding could be useful in tailoring to achieve desired viscosity to meet, for
15 example, particular coating equipment requirements.

TABLE 5

		Particle Size, micrometers, by volume			
“ZEEOSPHERES” Microspheres		Distribution			
Grade	Composition	10 th %	50 th %	90 th %	Effective top size
G-200	Silica-alumina ceramic	1	4	10	12
W-210	Alkali alumino-silicate ceramic	1	3	11	12
W-410	Alkali alumino-silicate ceramic	1	4	15	24
W-610	Alkali alumino-silicate ceramic	1	10	28	40

Example 3

The following compositions were prepared and tested according to the procedures of Example 1 and 2

5

TABLE 6

Component	Example 3.1	Example 3.2
"GENOMER 2258"	7.71	7.71
"GENOMER 2263"	18.00	18.00
"TMPTA"	25.71	25.71
"GENOCUR MBF"	0.65	0.65
"IRGACURE 819"	0.26	0.26
"BYK W-985"	0.43	0.43
"W210 ZEEOSPHERES"	47.08	47.08
"GF56"	0.00	0.59
"PARALOID 2655"	2.78	2.78

10 **TABLE 7**

	Example 3.1	Example 3.2
Viscosity, mPas	652	623
Modulus, MPa	6099	5678
Toughness, kgs ⁻² m ⁻¹	0.095	0.16

TABLE 8

Component	Example 3.3	Example 3.4
"GENOMER 2258"	3.48	3.45
"GENOMER 1343"	30.80	30.53
"TMPTA"	21.40	21.31
"GENOCUR MBF"	0.69	0.68
"SPEEDCURE BEM"	0.00	0.00
"IRGACURE 819"	0.28	0.28
"BYK W-985"	0.35	0.35
"W210 ZEEOSPHERES"	21.50	21.31
"GF56"	0.00	0.78
"SX400"	21.50	21.31

5 **TABLE 9**

	Example 3.3	Example 3.4
Viscosity, mPas @ 60° C	358	262
Modulus, MPa	6423	7270
Toughness, kgs ⁻² m ⁻¹	0.046	0.052

TABLE 10

Component	Example 3.5	Example 3.6	Example 3.7	Example 3.8
"SPEEDCURE 2263"	25.71	25.71	25.71	25.71
"TMPTA"	10.71	10.71	10.71	10.71
"SPEEDCURE 1223"	15.00	15.00	15.00	15.00
"BYK W-985"	0.43	0.43	0.43	0.43
"PARALOID 2655"	2.78	2.78	2.78	2.78
"W210 ZEEOSPHERES"	47.08	47.08	47.08	47.08
"GF31"	0.9	1.2	1.5	1.8
"GENOCUR MBF"	0.65	0.65	0.65	0.65
"IRGACURE 819"	0.26	0.26	0.26	0.26

TABLE 11

5

Component	Example 3.9	Example 3.10	Example 3.11	Example 3.12
"SPEEDCURE 2263"	25.71	25.71	25.71	25.71
"TMPTA"	10.71	10.71	10.71	10.71
"SPEEDCURE 1223"	15.00	15.00	15.00	15.00
"BYK W-985"	0.43	0.43	0.43	0.43
"PARALOID 2655"	2.78	2.78	2.78	2.78
"W210 ZEEOSPHERES"	47.08	47.08	47.08	47.08
"GF56"	0.69	0.92	1.15	1.38
"GENOCUR MBF"	0.65	0.65	0.65	0.65
IRGACURE 819	0.26	0.26	0.26	0.26

TABLE 12

"GF31" Methacryl functional silane				
Example	% of silane to filler	Viscosity, mPas	Mean Modulus, MPa	Mean Toughness, kgs ⁻² m ⁻¹
3.5	1.9	240	5879	0.139
3.6	2.5	241	5598	0.115
3.7	3.2	227	5421	0.173
3.8	3.8	248	5600	0.153

5 **TABLE 13**

"GF56" Vinyl functional silane				
Example	% of silane to filler	Viscosity, mPas	Mean Modulus, MPa	Mean Toughness, kgs ⁻² m ⁻¹
3.9	1.5	250	5472	0.132
3.10	2.0	220	5331	0.129
3.11	2.4	216	5380	0.106
3.12	2.9	166	4831	0.172

Example 4Example 4.15 **TABLE 14**

Component	
"TMPTA"	19.4
"ERL-4221"	45.3
"DAROCURE 1173"	0.6
"UVI-6974"	1.9
"BYK W-985"	0.4
"SX400"	32.3

This example was prepared and poured into a mould as in Example 1. The example was cured by passing through the "MINICURE" UV curing unit twice.

10 The front side was cured but the back was still liquid.

Example 4.2**TABLE 15**

15

Component	
"TMPTA"	19.8
"ERL-4221"	46.3
"DAROCURE 1173"	0.6
"IRGACURE 819"	0.1
"UVI-6974"	2.0
"BYK W-985"	0.3
"SX400"	30.9

This example was prepared and poured into a mould as in Example 1. The example was cured by passing through the "MINICURE" UV curing unit twice.

The front and back sides were both cured demonstrating the advantage of the visible light curing action of Irgacure 819 with SX400 filled examples.

In a further experiment, Speedcure TPO was found to be another suitable visible light catalyst.

TABLE 16

Component	Example 4.3	Example 4.4
"GENOMER 2263"	19.50	19.50
"TMPTA"	27.24	27.24
"BYK W-985"	4.22	4.22
"GENOCUR MBF"	0.31	0.31
"IRGACURE 819"	0.24	-
"SPEEDCURE TPO"	-	0.24
"SPEEDCURE PBZ"	0.52	0.52
"SPEEDCURE DMB"	0.59	0.59
"ZEEOSPHERES W-210"	31.34	31.34
"SX-400"	11.48	11.48
GF56	0.59	0.59
"PARALOID 2655"	2.53	2.53
"GENORAD 10"	1.08	1.08

10 **TABLE 17**

	Example 4.3	Example 4.4
Modulus, Mpa	6572	6599
Toughness, kgs-2m ⁻¹	0.07	0.07

Example 5**TABLE 18**

Component	Example 5.1	Example 5.2	Example 5.3
GENOMER 2263	20.0	20.0	20.0
"TMPTA"	19.1	19.1	19.4
"HDDA"	12.0	12.0	12.0
"GENOCUR MBF"	0.8	0.8	0.8
"IRGACURE 819"	0.3	0.3	0.3
"BYK W-985"	0.2	0.2	0.2
"W210 ZEEOSPHERES"	44.3	34.2	26.8
"SX-400"	0.0	10.14	17.54
"GF56"	0.56	0.61	0.64
"PARALOID 2655"	3.29	3.29	3.29

5

Increased levels of mica increase flexural modulus and decrease toughness.

	Example 5.1	Example 5.2	Example 5.3
Modulus, Mpa	5724	6765	7527
Toughness, kgs-2m ⁻¹	0.13	0.09	0.06

10 Example 6

The following coating composition was prepared and tested as in Example 1 and 2:

Example 6.1

TABLE 19

"TMPTA"	25.78
"GENOMER 2263"	25.78
"GENOCURE MBF"	0.66
"IRGACURE 819"	0.27
"BYK W-985"	0.44
"W-210 ZEEOSPHERES"	47.08

5

The cured composition had a flexural modulus of 10340 MPa and toughness 0.039 kgs⁻²m⁻¹.

An abrasive material was prepared comprising a 3M manufactured 2.97

- 10 thousands of an inch thick (75.4 micrometers) polyester backing, 20 g/m² UV cured hot-melt polyester make adhesive, 180 g/m² abrasive grains comprising 30% of a sol-gel 80 grit alumina abrasive grain product, marketed by the 3M Company under the trade designation "CUBITRON 222" and 70% "P.80" blue aluminium oxide (Treibacher BFRPL), Example 7.1 as the size at
- 15 140 g/m², and a conventional calcium stearate/styrene acrylate binder supersize at a 11g/m².dry coating weight. The make and supersize formulations were identical to those used on the coated abrasive product marketed by the 3M Company under the trade designation 3M 255P STIKIT™ commercially available from 3M Company.

20

The abrasive material was converted into self-adhesive discs and tested on a random orbital sander sanding body-filler, using "Test Method for Cut" described below, and its performance compared with a coated abrasive product marketed by the 3M Company under the trade designation "255P STIKIT"™. A 10% improvement in cut over the commercial product was

25 observed.

Test Method for Cut

a) Filler Panel Preparation

- 5 The polyester body filler used was "STANDOX POLYESTER, EXPRESS PLASTIC, EXPRESS BODY FILLER" and "STANDOX HARDENER PASTE" available from Standox UK, Du Pont Performance Coatings (UK) Ltd., Freshwater Road, Dagenham, Essex, RM8 1RU, UK. The filler was stored at 20°C for 24 hours before testing. The ratio used was 800 parts by weight of
10 filler to 24 parts by weight of hardener. The total amount of filler needed for all the test program was blended using a pneumatic stirrer to mix the filler until it was of a consistent viscosity with all lumps removed.

b) Mixing

- 15 According to the polyester filler mixing instructions, the requisite ratio of polyester filler and hardener paste was poured into a beaker. This was mixed well ensuring an even distribution of the hardener, and poured directly into a clean mould, spreading it evenly in the mould in order to produce a slab of uniform thickness. The mould was vibrated by tapping it on the work
20 surface, to bring any air bubbles to the surface. The time of mixing was written on the side of the aluminium plate. At approximately 40 minutes the mould was removed after cutting through the double sided tape on the inside of the mould.

25 c) Testing

1 hour after mixing the filler panel were ready for testing. It had a test life of 30 minutes, so the panels needed to be mixed to produce a new one every 30 minutes.

30 d) Sander

- A 10mm Desoutter random orbital pneumatic sander, model number 567. available from Desoutter Sales Ltd, Eaton Road, Hemel Hempstead, Hertfordshire HP2 7DR was set up with a hard foam backup pad. Several
35 discs were used for each test lot. Each panel was weighed before and after

sanding for 75 and 135 seconds. The weight loss of a panel was the measure of "cut".

Example 7

Formulations listed in Table 20 and 22 were compared with the size used in a standard conventional abrasive, a coated abrasive product marketed by the 3M Company under the trade designation "3M 255P STIKIT"TM, a typical urea formaldehyde size construction.

The following coating compositions were prepared and tested as in Example 1 and 2:

TABLE 20

Component	Example 7.1	Example 7.2	Example 7.3	Example 7.4	Example 7.5	Example 7.6
"GENOMER 2263"	22.83	22.41	8.24	2.21	5.37	16.15
"TMPTA"	19.86	11.99	25.15	28.33	13.38	11.83
"GENOMER 1223"	21.39	25.61	20.46	20.02	25.26	22.33
"GENOCUR MBF"	0.81	0.76	0.68	0.64	0.56	0.64
"IRGACURE 819"	0.33	0.30	0.27	0.26	0.22	0.26
"BYK W-985"	0.21	0.24	0.27	0.33	0.35	0.31
"W-210 ZEEOSPHERES"	26.86	27.82	37.33	29.30	40.09	35.44
"SX400"	4.70	7.78	4.47	15.68	11.29	9.98
"GF56"	0.42	0.48	0.54	0.63	0.70	0.61
"PARALOID 2655"	2.59	2.59	2.59	2.59	2.78	2.45

TABLE 21

	Example 7.1	Example 7.2	Example 7.3	Example 7.4	Example 7.5	Example 7.6
Modulus, Mpa	4377	5000	6000	8005	7942	5771
Toughness, kgs ⁻² m ⁻¹	0.13	0.15	0.12	0.06	0.03	0.17

TABLE 22

5

Component	Example 7.7	Example 7.8	Example 7.9	Example 7.10	Example 7.11	Example 7.12
"GENOMER 2263"	14.77	25.07	27.35	19.87	25.24	21.83
"TMPTA"	17.03	23.31	21.00	27.76	25.14	22.64
"GENOMER 1223"	22.59	8.92	6.19	4.30	0	3.95
"GENOCUR MBF"	0.69	0.73	0.69	0.69	0.64	0.61
"IRGACURE 819"	0.28	0.29	0.28	0.24	0.26	0.25
"BYK W-985"	0.29	0.25	0.28	0.32	0.31	0.32
"W-210 ZEEOSPHERES"	32.41	32.19	31.80	31.94	35.38	36.84
"SX400"	9.12	5.63	8.90	11.70	9.96	10.37
"GF56"	0.56	0.50	0.55	0.60	0.61	0.64
"PARALOID 2655"	2.25	3.10	2.96	2.58	2.45	2.55
Total	100	100	100	100	100	100

TABLE 23

	Example 7.7	Example 7.8	Example 7.9	Example 7.10	Example 7.11	Example 7.12
Modulus, Mpa	4217	1581	2661	7335	3704	2870
Toughness, kgs ⁻² m ⁻¹	0.33	0.78	0.83	0.07	0.09	0.07

The comparable Modulus and Toughness properties for the standard product, a coated abrasive product marketed by the 3M Company under the trade designation "3M 255P STIKIT"TM were 6536 MPa and $0.05 \text{ kgs}^{-2}\text{m}^{-1}$ respectively

- 5 Various modifications and alterations of this invention will become apparent to those skilled in the art without departing from the scope and spirit of this invention, and it should be understood that this invention is not to be unduly limited to the illustrative embodiments set herein.

CLAIMS

1. An abrasive article comprising abrasive particles and binder material, wherein the binder material comprises a UV- cured formulation and aluminosilicate ceramic microspheres substantially transparent to UV-
5 radiation, wherein the aluminosilicate ceramic microspheres are present in a range from 20 to 80 percent by weight, based on the combined weight of the cured formulation and the aluminosilicate ceramic microspheres, wherein the aluminosilicate ceramic microspheres have an average particle size in a range from 1 micrometer to 40 micrometers.
10
2. An abrasive article as claimed in Claim 1 which comprises from 40 to 60% by weight of microspheres, based on the combined weight of said formulation and said microspheres.
- 15 3. An abrasive article as claimed in Claim 1 or Claim 2 in which the microspheres have an average particle size in a range from 1 micrometer to 20 micrometers.
4. An abrasive article as claimed in Claim 3 in which the microspheres
20 have an average particle size in a range from 1 micrometer to 10 micrometers.
5. An abrasive article as claimed in any preceding claim in which the microspheres have an average particle size about 3 micrometers.
25
6. An abrasive article as claimed in any preceding claim in which the microspheres are transparent to visible light.
7. An abrasive article as claimed in any preceding claim in which said
30 formulation further comprises mica and a visible light activated catalyst.
8. An abrasive article as claimed in Claim 7 in which the mica is present in an amount up to at least 22% by weight of the combined weight of said formulation, microspheres and mica.

9. An abrasive article as claimed in Claim 7 or Claim 8 in which, by weight, more of the microspheres are present than are the mica.

5 10. An abrasive article as claimed in any preceding claim in which said formulation further comprises up to 2% by weight of an organo functional silane, based on the weight of filler.

11. An abrasive article as claimed in Claim 10 in which the organo
10 functional silane is a vinyl functional silane.

12. An abrasive article as claimed in any preceding claim in which said formulation additionally comprises up to 2% by weight of a surfactant.

15 13. An abrasive article as claimed in any preceding claim in which the said formulation is obtained by curing a composition comprising epoxy acrylate.

14. An abrasive article as claimed in any preceding claim comprising a layer of said binder material.

20

15. An abrasive article as claimed in Claim 14 in the form of a coated abrasive comprising an abrasive layer secured to a major surface of a backing, the abrasive layer comprising abrasive particles and said binder material.

25

16. An abrasive article as claimed in Claim 15 wherein said abrasive layer comprises a make coating and a size coating, and wherein at least one of said make and size coating comprises said binder material.

30 17. An abrasive article as claimed in Claim 15 in which the size coating comprises said binder material.

